

Nanotube-Based Structures for Superstrong Materials, Nanoscale Sensors and Devices, and Efficient Electron Emitters

**J. Bernholc, M. Buongiorno Nardelli, V. Meunier,
S. Nakhmanson, C. Roland, and Q. Zhao**

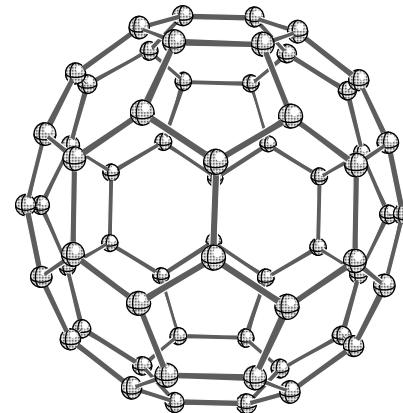
North Carolina State U., Raleigh, NC 27695-8202

- I. Introduction to nanotubes**
- II. Mechanical properties: the strongest materials known**
- III. Energy storage: high capacity, fast charging Li/NT batteries**
- IV. Pyro- and piezo-electric effects in BN/C nanotubes
Composite BN/C nanotubes – field emission**
- V. Nanotube-metal clusters sensor structures
single molecule sensitivity**
- VI. Summary**

Early simulation of C₆₀ solid

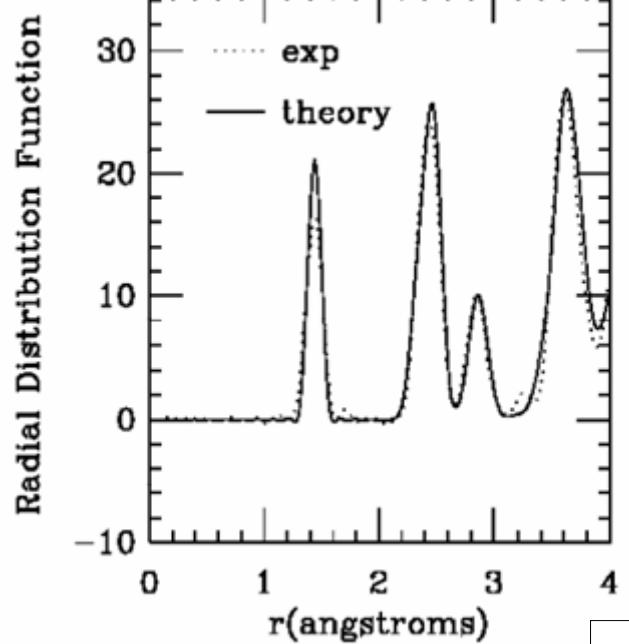
**solid C₆₀ geometry optimization
no symmetry constraints
fcc structure, a=14.2 Å (from exp.)**

	Double	Single
CP (35 Ry)	1.40	1.45
CP (26 Ry)	1.41	1.45
exp (NMR)	1.40	1.45

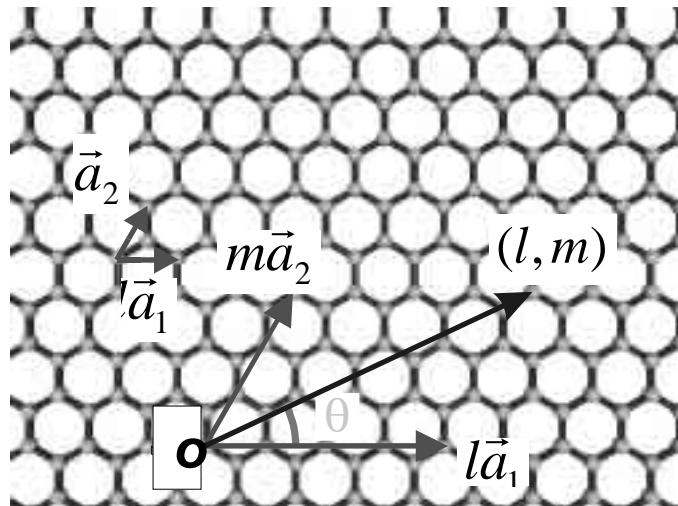


Zhang, Yi,
Bernholc, PRL 91

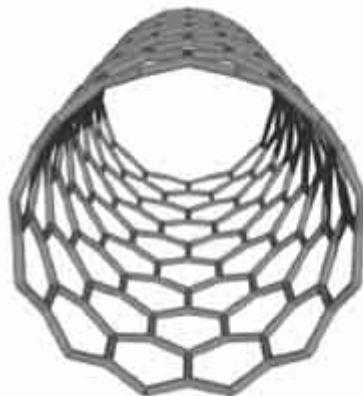
**excellent agreement with photoemission
(Weaver et al), EXAFS (Cox et al), NMR,
and also with subsequent neutron diffraction data
(Li, Lannin, et al)**



Introduction to nanotubes



	Properties	(l, m) relation
Geometric	radius	$R = \frac{\sqrt{3}d\sqrt{l^2 + m^2 + lm}}{2\pi}$
	chiral angle	$\theta = \arcsin \frac{\sqrt{3}m}{3\sqrt{l^2 + m^2 + lm}}$
Electronic	metal	$\text{mod}(l - m) = 3$
	semi-conductor	$\text{mod}(l - m) \neq 3$



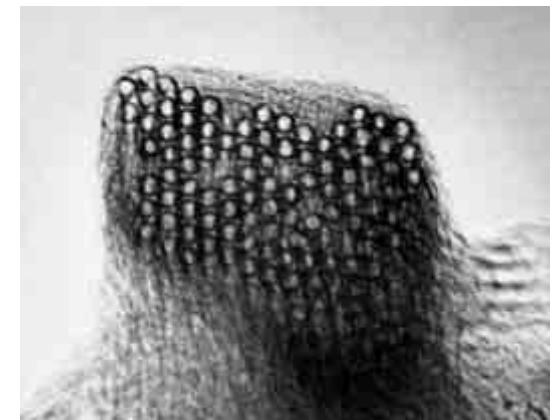
(12,0)
zigzag



(6,6)
armchair



(6,4)
chiral

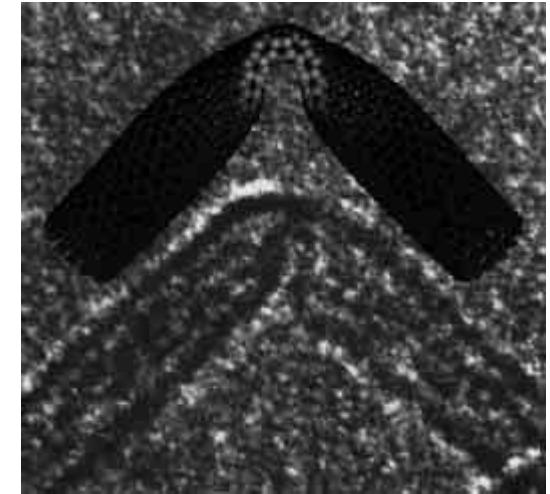
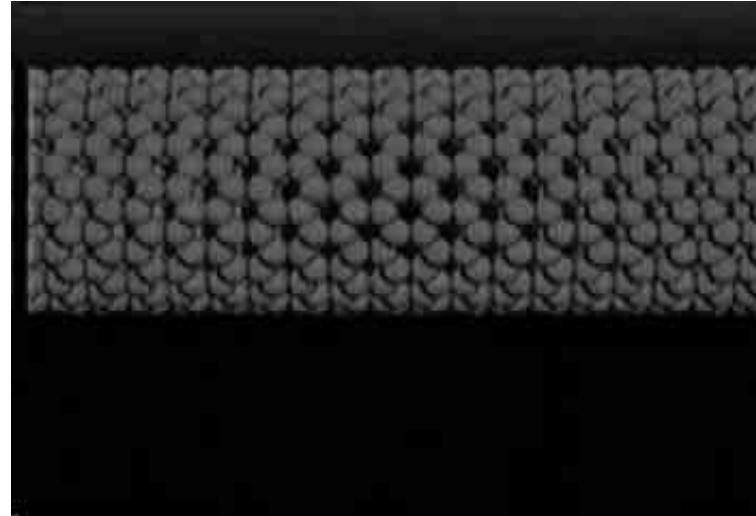


nanotube “rope”
Thess, Smalley, et. al., Science 96

Elasticity and Breakage of Carbon Nanotubes

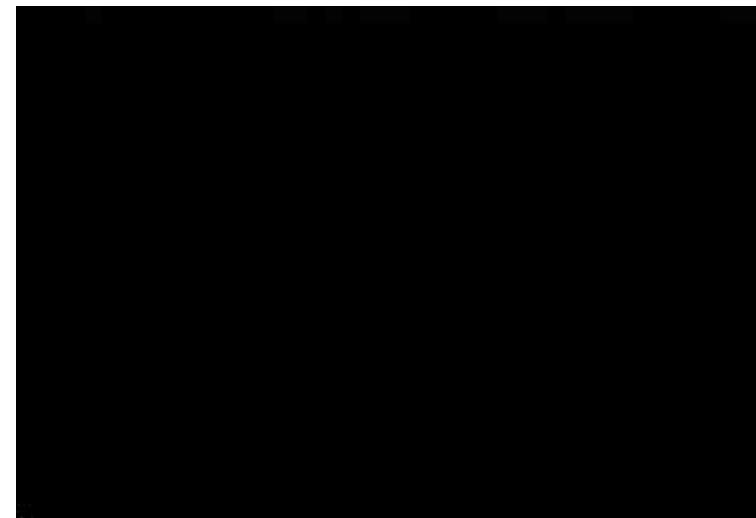
Bending

Iijima, Brabec,
Maiti, Bernholc
JCP **104**, 2089 (1996)

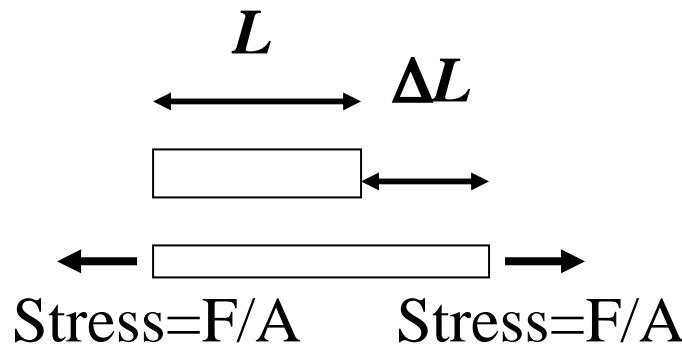


Compression

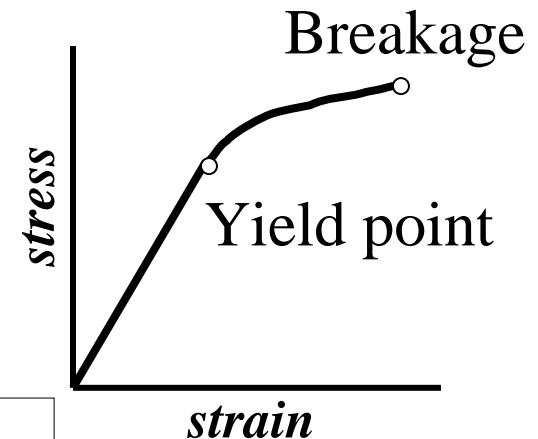
Yakobson, Brabec, Bernholc
PRL **76**, 2511 (1996)



Introduction to strength of materials



$$\frac{F}{A} = Y \frac{\Delta L}{L}$$

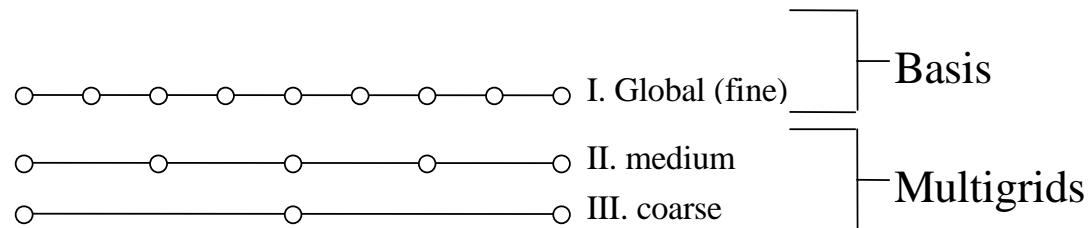


Young's modulus determines stiffness

Material	Y [GPa]	Yield point [GPa]	Max $\Delta L/L [\%]$
Graphite	1080	31.3	2.9
Steel	220	4.2	1.9
Glass	168	3.6	2.2
Silicon	163	4.1	2.5

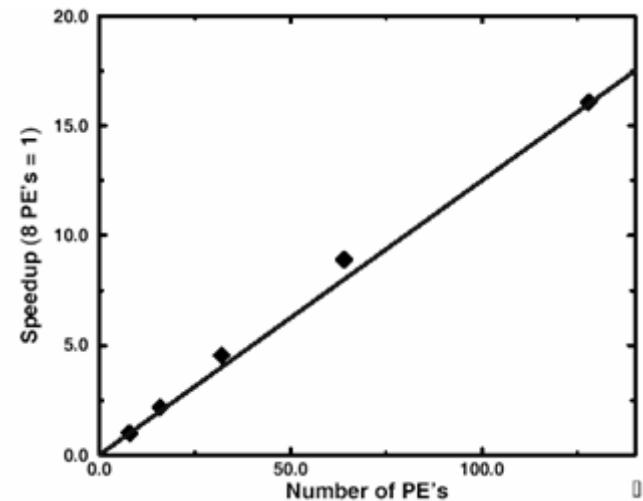
Most materials can be stretched by at most 3%.

Multigrid method for quantum simulations



- Density functional equations solved directly on the grid
- Multigrid techniques remove instabilities by working on one length scale at a time
- Convergence acceleration and automatic preconditioning on all length scales
- Non-periodic boundary conditions are as easy as periodic
- Compact “Mehrstellen” discretization
- Allows for efficient massively parallel implementation

See E. L. Briggs, D. J. Sullivan and J. Bernholc *Phys. Rev. B* **54**, 14362 (96).

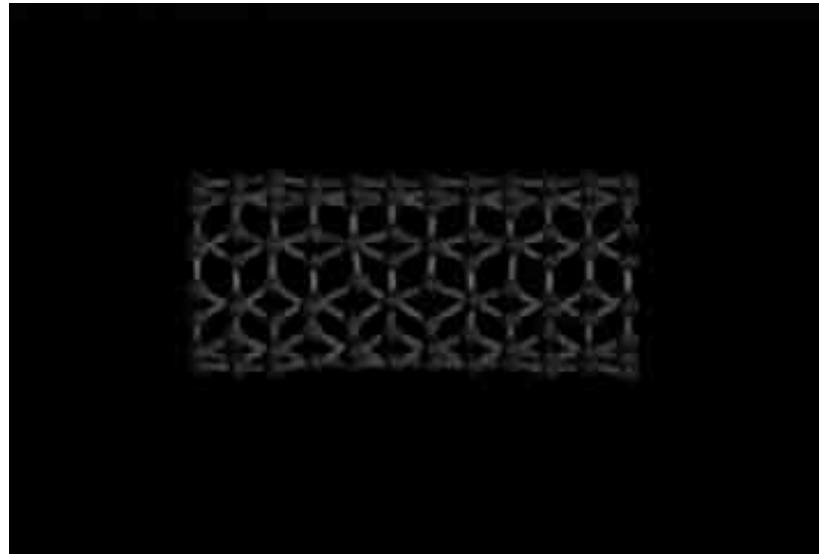
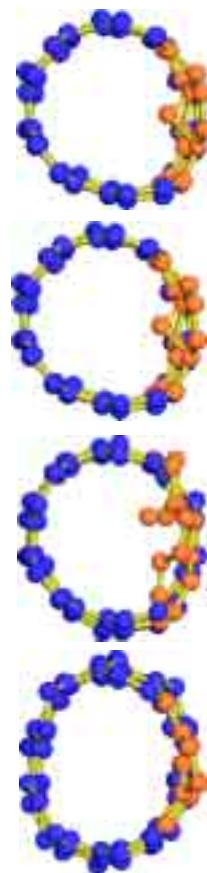
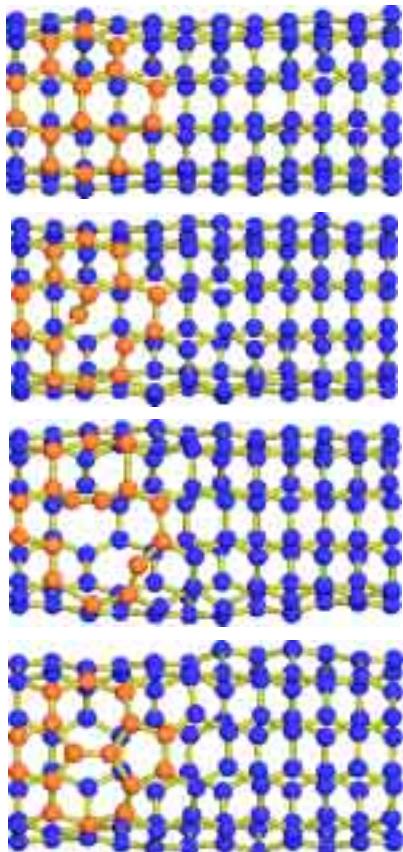


Speedup on Cray T3E with number of processors

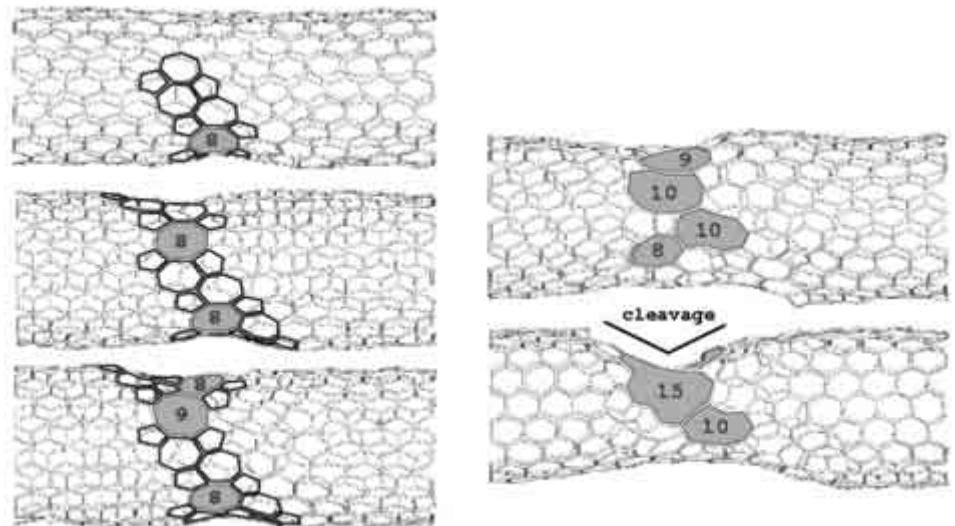
Runs also on IBM SP, Origin 2000 and Linux clusters

Breakage of nanotubes

Nanotubes break by first forming a bond rotation 5-7-7-5 defect.

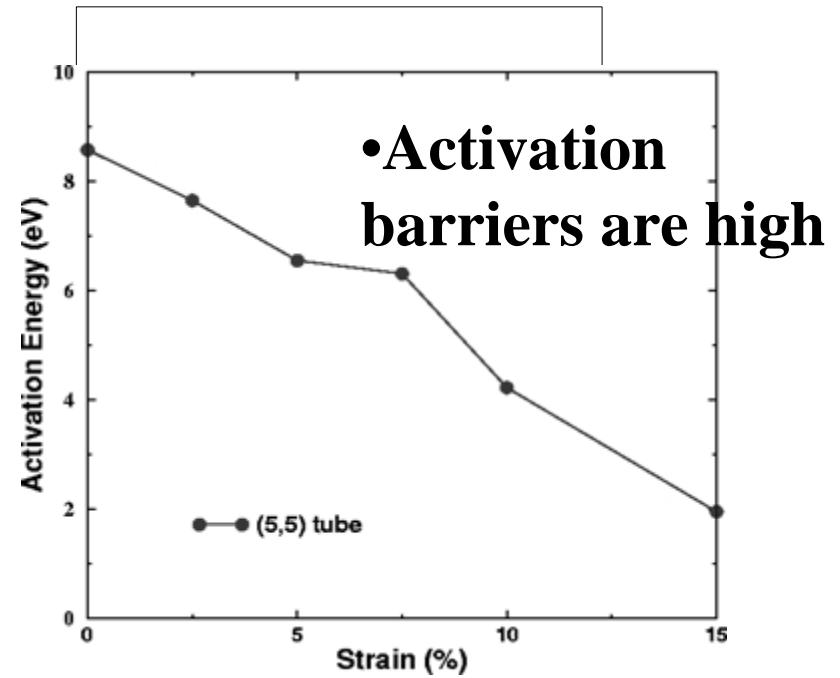
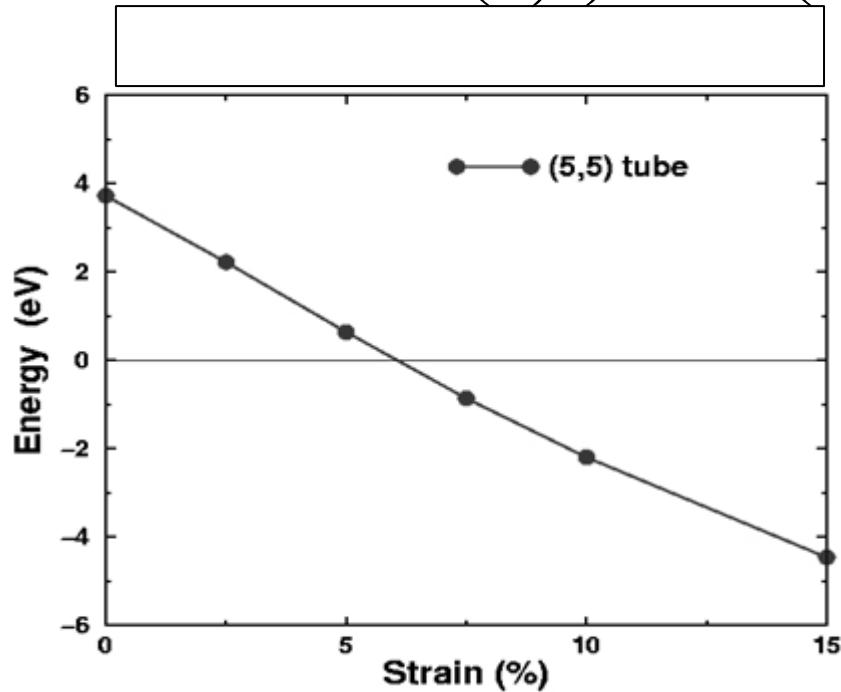


Additional bond rotations lead to larger defects and cleavage.



Buongiorno Nardelli, Yakobson, Bernholc
PRB 57, R4277 (1998).

Formation and activation energies of 5775 defect in (5,5) tube (*ab initio* results)

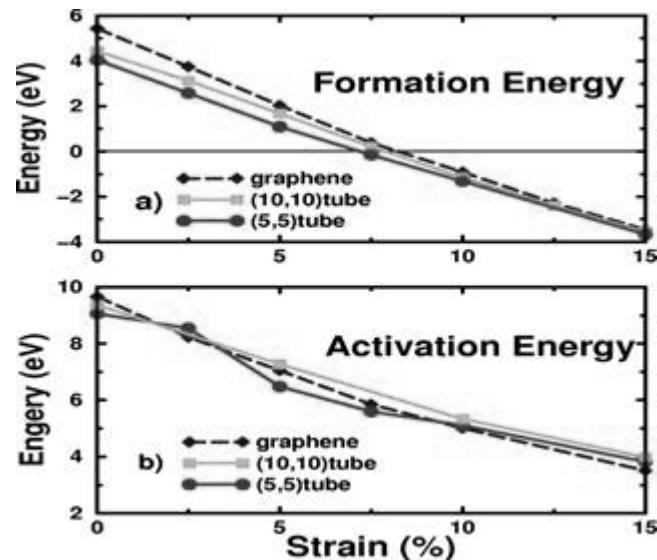


Formation and activation energies decrease with strain

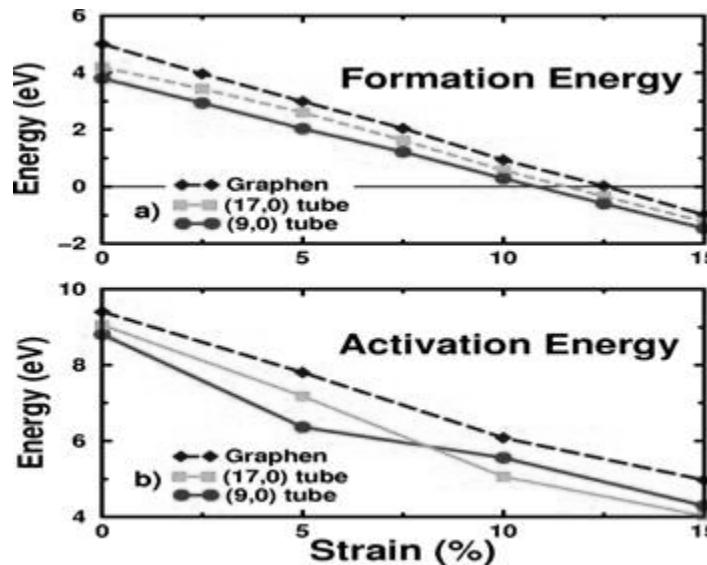
- 5775 defect is favored for strains above $\sim 6\%$
- Experimentally, tubes break at around 5-6% strain
(Walters, Smalley et al APL, 1999; Yu, Ruoff et al PRL, 2000)

Curvature and chirality effect on defect formation

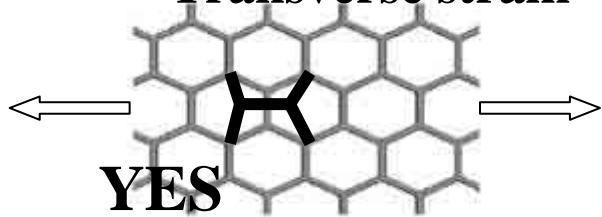
armchair tubes



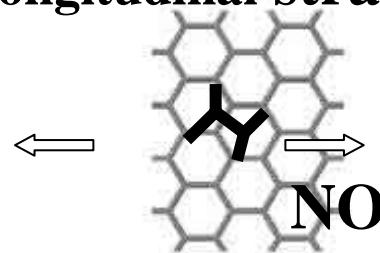
zigzag tubes



Transverse strain



Longitudinal strain



In armchair NT the 5775 energetically favorable at 5-6%

In zigzag NT the 5775 energetically favorable at 10-11%

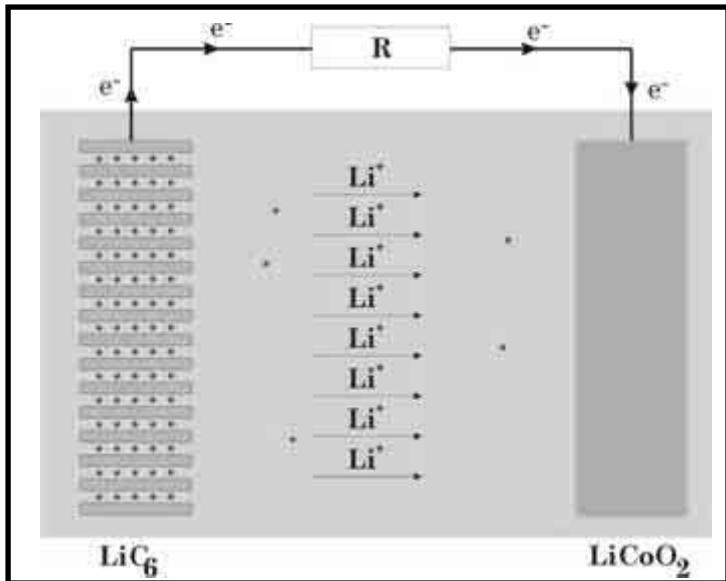
Activation energies are similar

Zhao, Buongiorno Nardelli, Bernholc PRB 65, 144105 (2002)

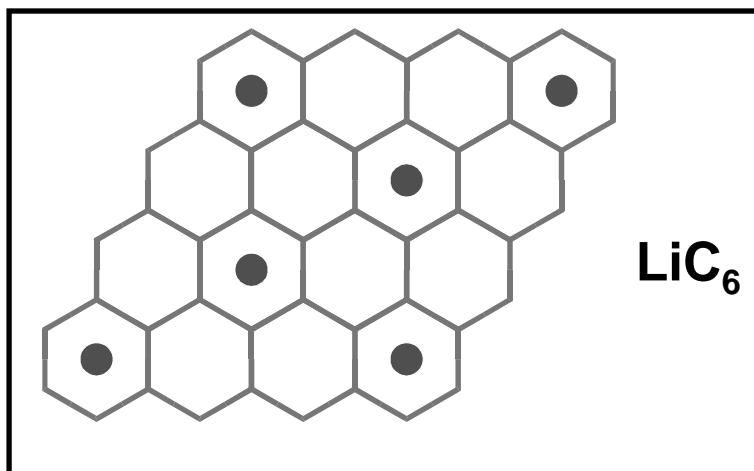
See also: Buongiorno Nardelli, Yakobson, Bernholc PRL 81, 4656 (1998)

Zhang, Lammert, Crespi, PRL 81, 5346 (1998)

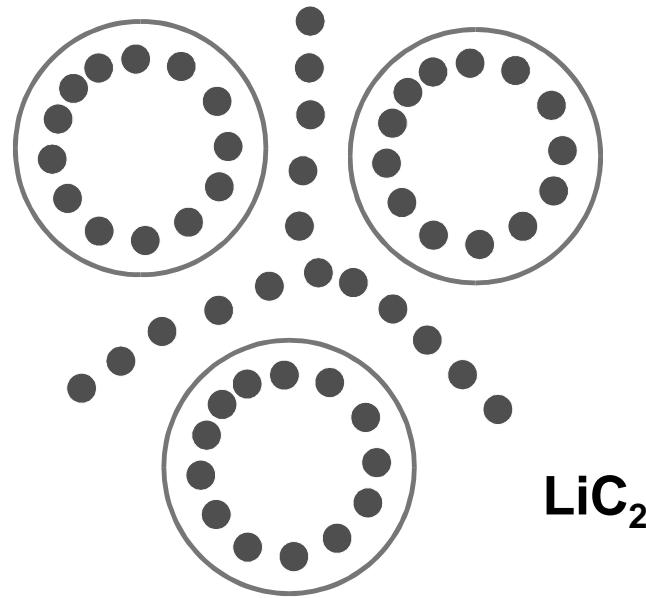
Li/Carbon-based high performance batteries



Lithium “rocking chair” battery

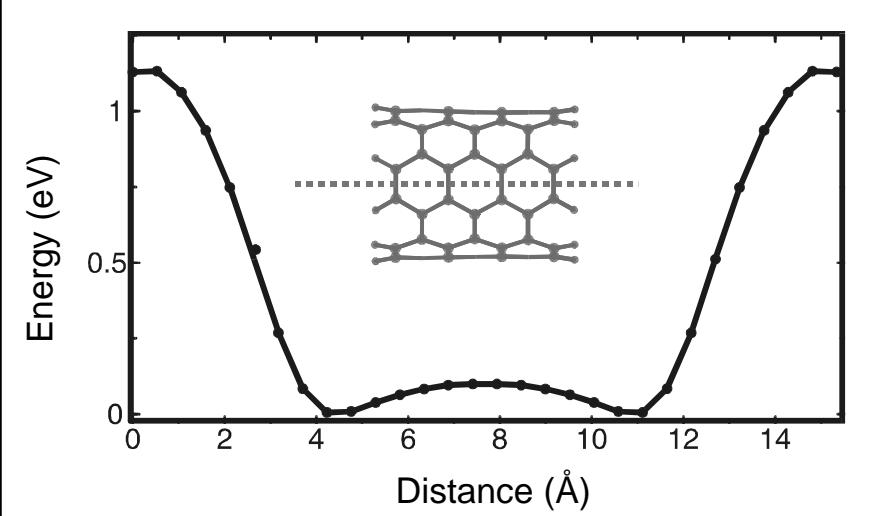


Nanotube-based anode



Two channels

Diffusion through open tip

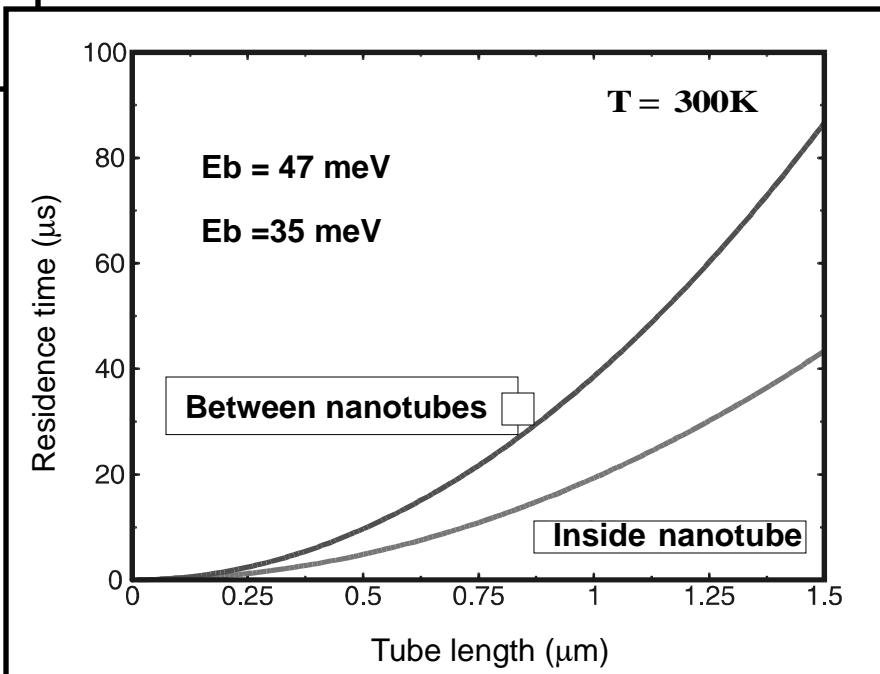


Li enters freely inside
H-saturated open nanotube

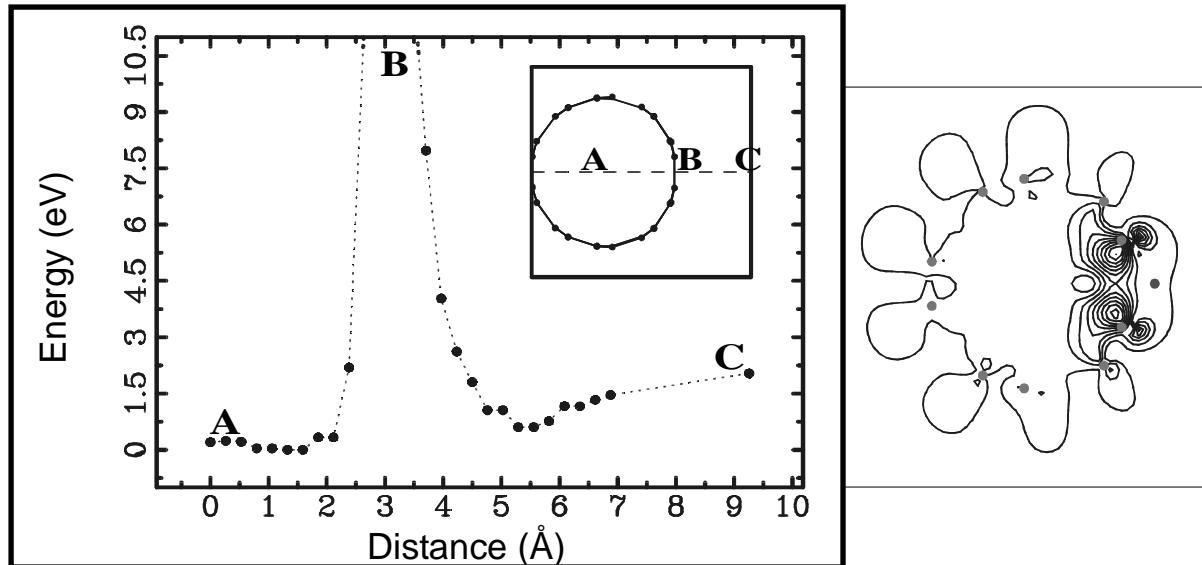
Once inside, the ions are not diffusion limited, provided that the tubes are relatively short

$$1\text{D motion: } \tau = \frac{L^2}{v a_0^2} e^{E_b / k_b T}$$

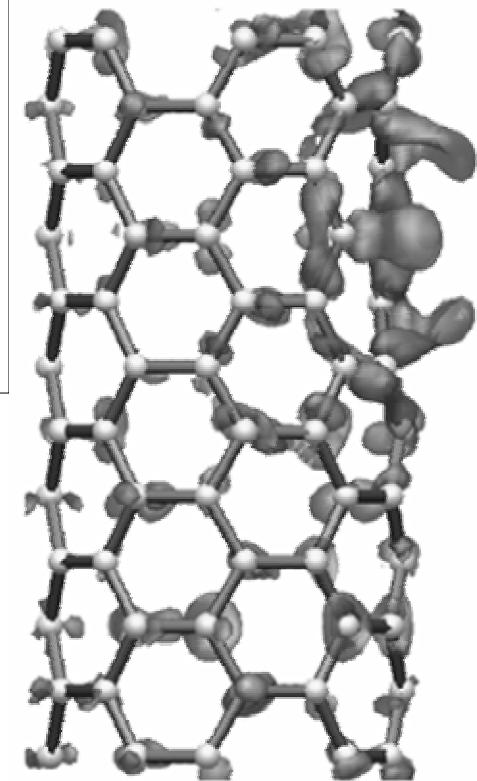
$v=2.10^{12}\text{hz}$



Diffusion through side-walls: *pristine* case

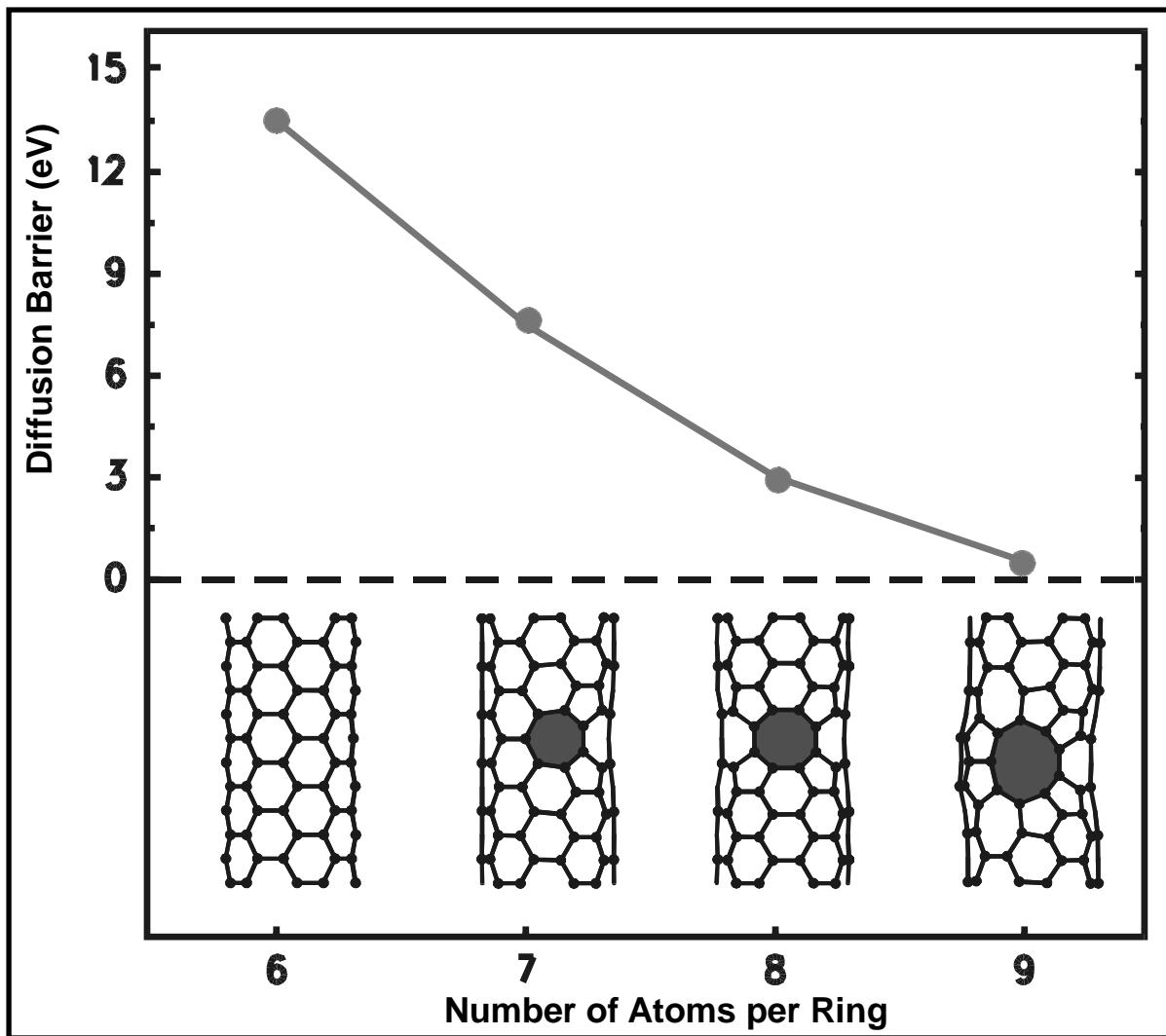


**Li does NOT penetrate
through an unaltered
nanotube**



Charge transfer

SW-types : heptagon, octagon, enneagon

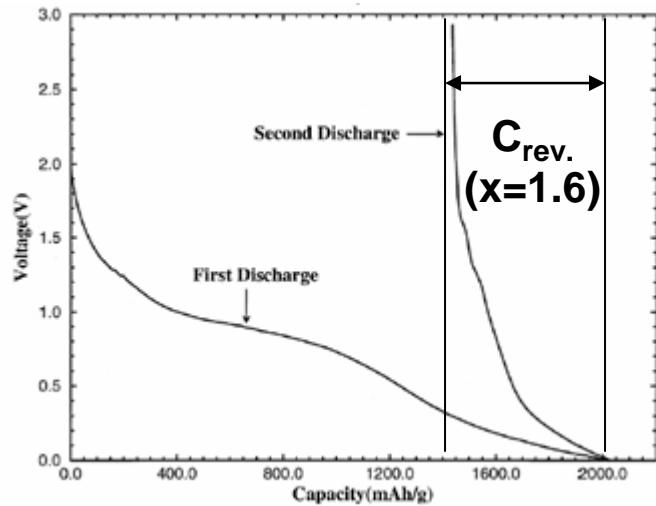


Li can freely enter through 9- (or higher) order topological defects

V. Meunier, C. Roland, J. Bernholc, Phys. Rev. Lett. **88**, 075506 (2002).

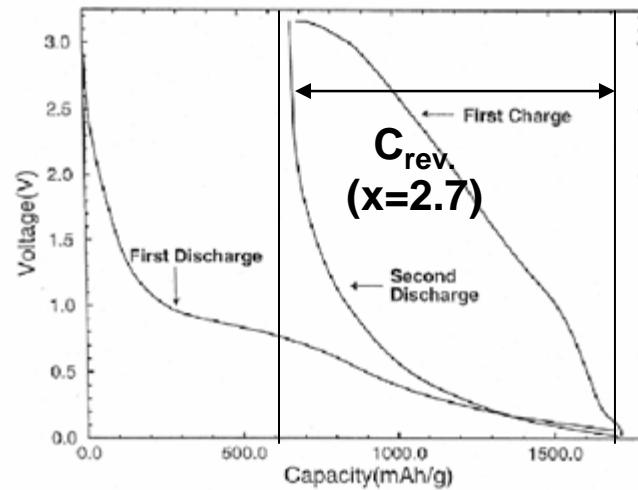
Using graphitic material as an anode host in Li_xC_6

Before ball-milling:



Gao, et al., CPL 307, 153 (1999).

After ball-milling:

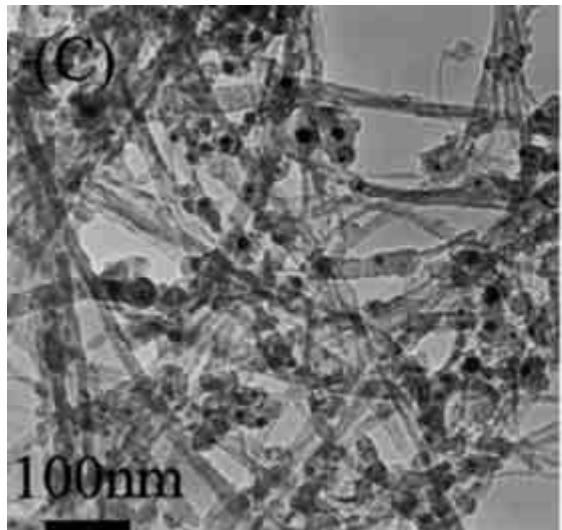


Gao, et al., CPL 327, 69 (2000).

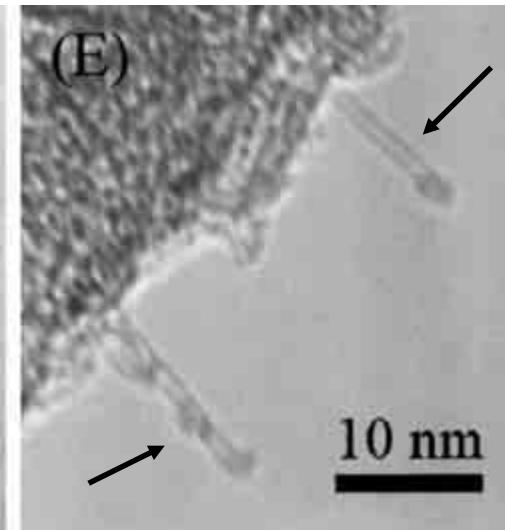
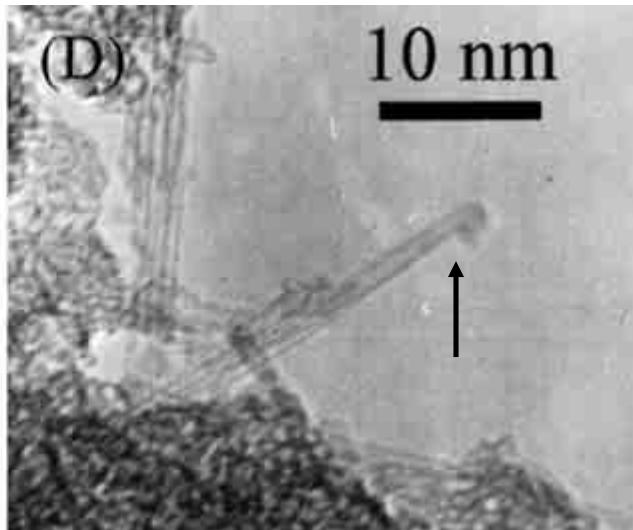
How can Li access all the intercalating sites?

Effect of the ball-milling on the nanotube material

Before ball-milling:



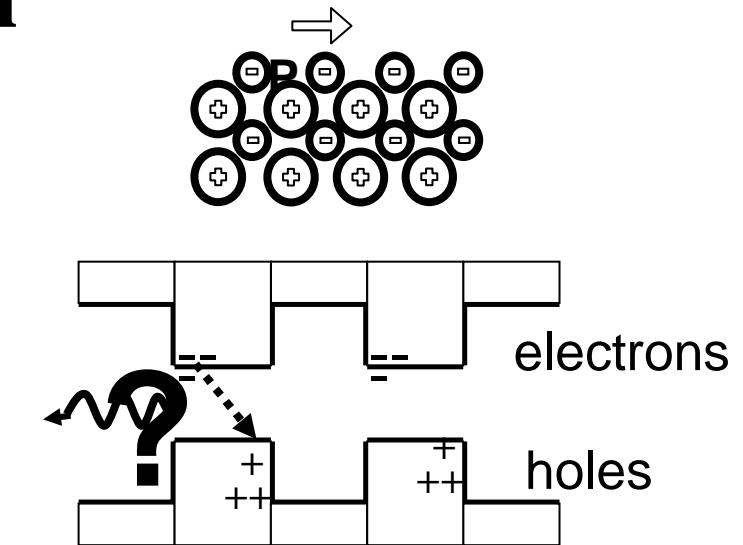
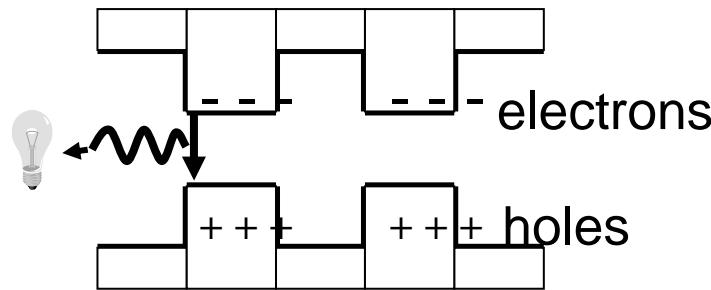
After 10min ball-milling:



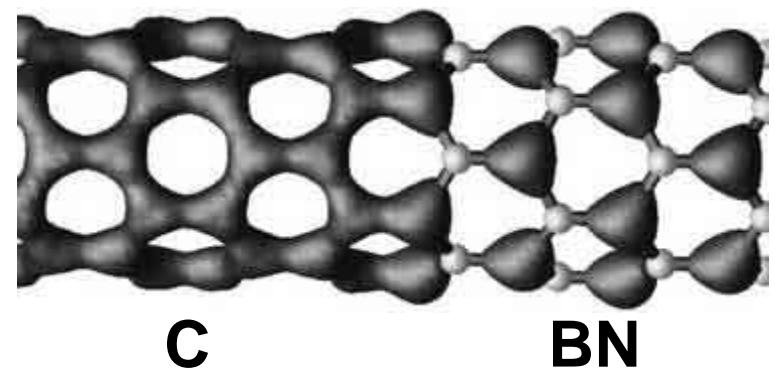
-Observation of nanotube ends: the SWNT's were fractured and shortened by BM.

-Further BM decreases the reversible capacity: increase of the amorphous carbon.

Symmetry and polarization



- BN nanotubes have sufficiently low symmetry to exhibit significant macroscopic polarization inside the nanotube.
- The resulting electric field can separate electrons and holes in quantum wells, inhibiting recombination.
- Nanoscale pyro- and piezoelectrics are useful: strain sensors, actuators, transducers



Field emission

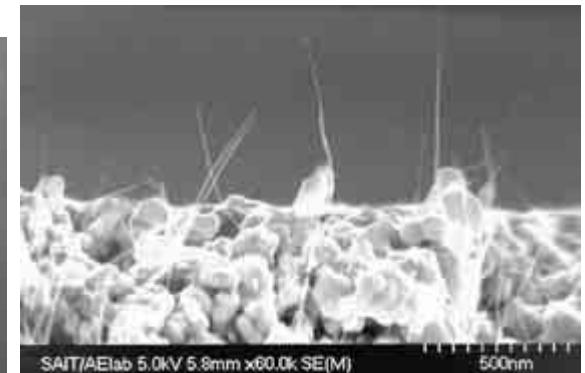
- Possible field emission enhancement from local polarization fields
- Fowler-Nordheim formula

$$J = \frac{C}{\phi} \beta E^2 e^{\frac{-B\phi^{3/2}}{\beta E}}$$

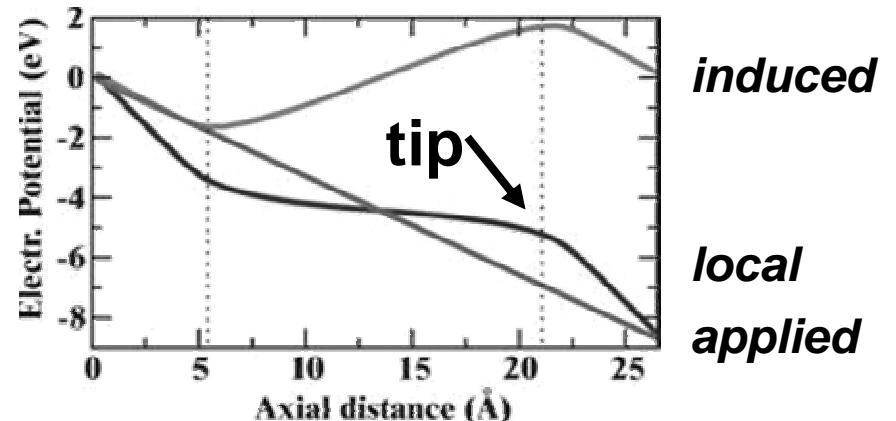
- E - applied electric field
- Φ - work function of the metal
- β - field enhancement factor

- Field enhancement not changed with the addition of BN sections

$$E_{\text{induced}}/E_{\text{applied}} \sim 2.1$$

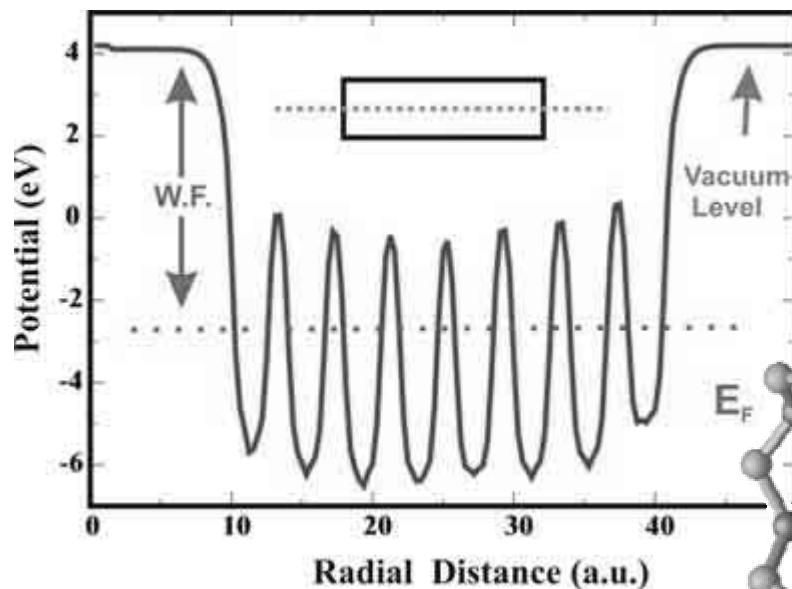


(Choi et al. APL (1999))

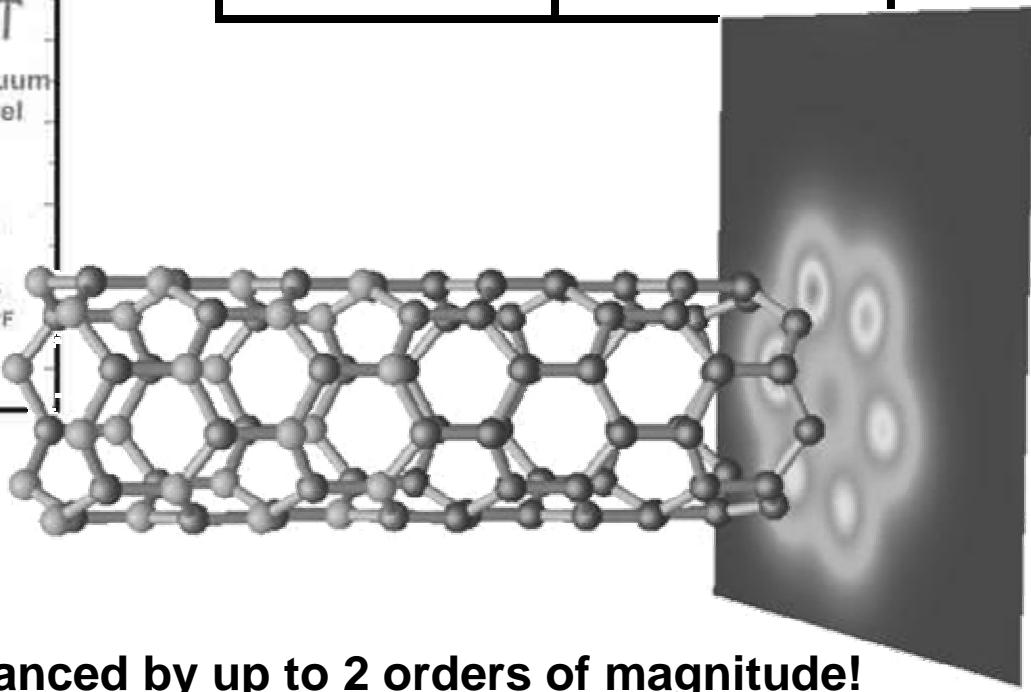


Work Function

- Work function at the tip of C(6,0)
- Dipole correction in the vacuum



BN/C	6.71 (eV)
C	6.41
NB/C	5.04



Field emission properties enhanced by up to 2 orders of magnitude!

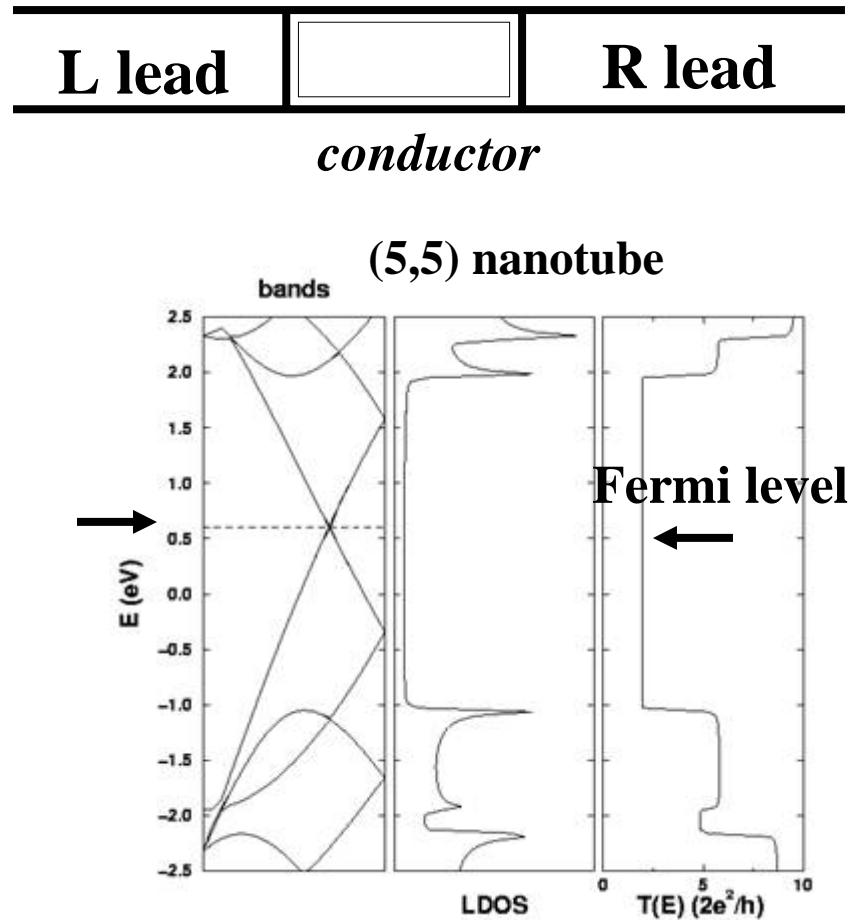
Meunier, Roland, Bernholc, Buongiorno Nardelli, APL (2002)

Introduction to quantum conductance

- Electron transmission through a device requires a Green's function treatment of an *open* system

Left lead -- Conductor -- Right Lead

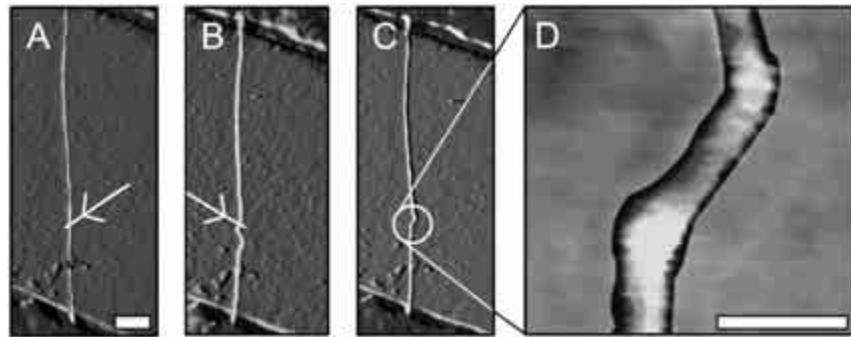
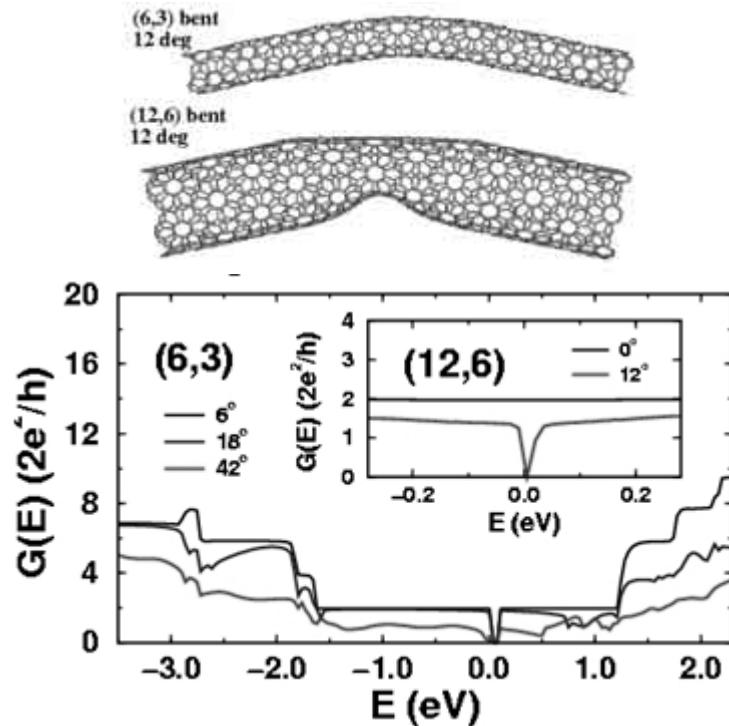
- In general, the quantum conductance measures the number of electron channels extending through the conductor *and* the leads, each contributing $2e^2/h$.
- For a perfect metallic nanotube and perfect contacts, both bands at the Fermi level contribute equally.
- For a disordered nanotube or for poor contacts, the conductance is much less.
- Conductances computed using a new, very efficient method (Buongiorno Nardelli PRB 1999; Buongiorno Nardelli and Bernholc, PRB RC 1999, Buongiorno Nardelli, Fattebert, Bernholc, PRB 2001)



Two bands cross at the Fermi level
⇒ Conductance ≡ $T(E_F) = 2$
Units of $2e^2/h \approx (12.9 \text{ k}\Omega)^{-1}$

Bent nanotubes and single electron transistors

Large diameter nanotubes are metallic when $n-m=3k$, but bending opens a sizable pseudo-gap.



Kinking nanotubes leads to single electron transistors operating at room T Postma, Dekker et al, Science 2001

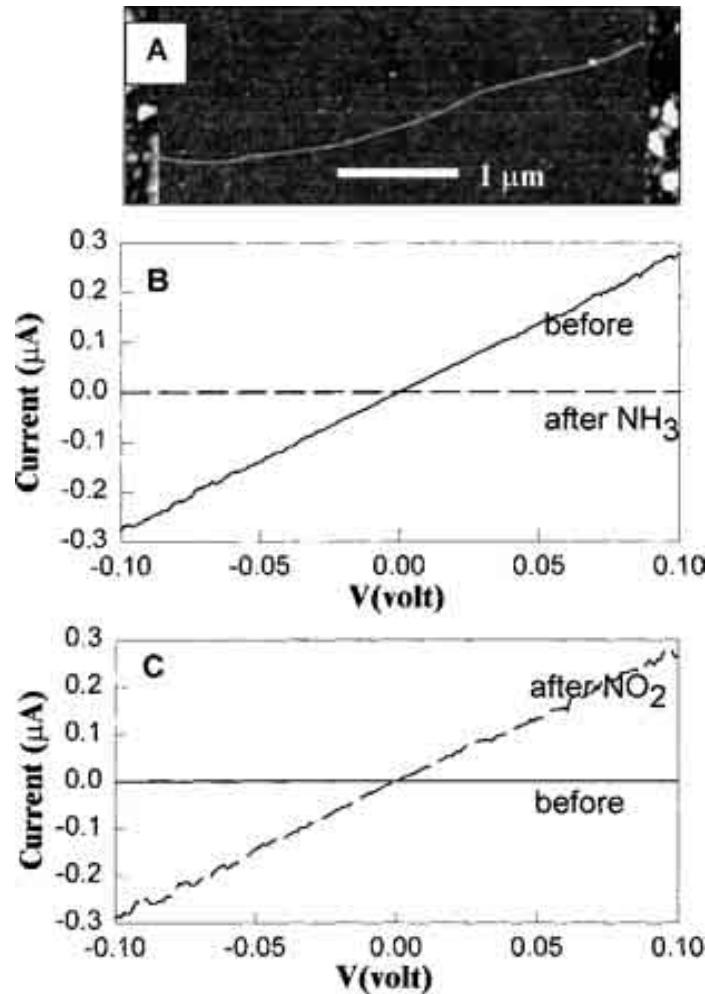
- * nanostrain sensors → chiral metallic NT
 - * flexible ballistic conductors → armchair NT
 - high bending angles break conductance
- Buongiorno Nardelli & J. Bernholc PRB RC (1999)

Carbon nanotubes as chemical sensors

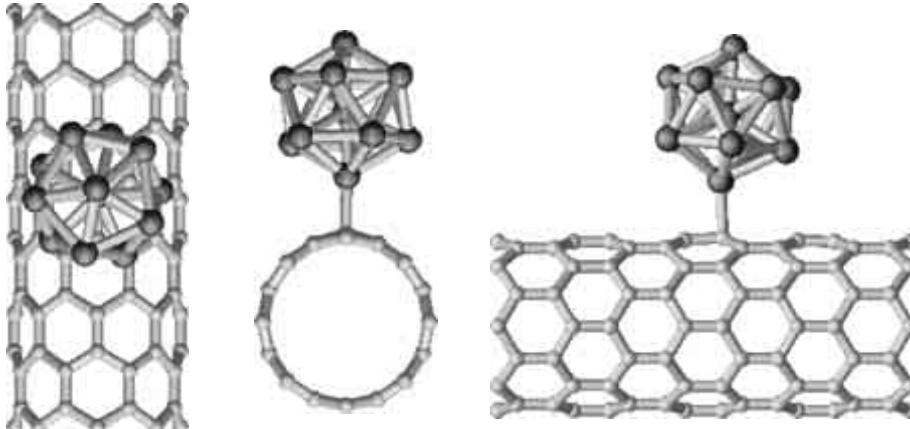
Adsorption of chemical species on single wall nanotubes can induce changes in the electrical characteristics

(Kong *et al.*, *Science* 2000)

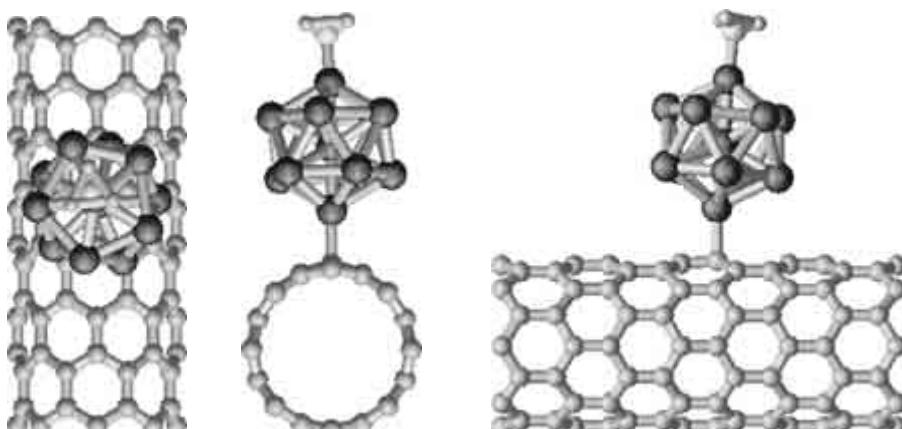
Semiconducting SW-CNT before and after exposure to NO_2 and NH_3 :
 NO_2 binds to the SWNT
 NH_3 does not bind directly:
- gating effect through substrate
- binding to adsorbed species



Semiconducting SWNT-metal assemblies



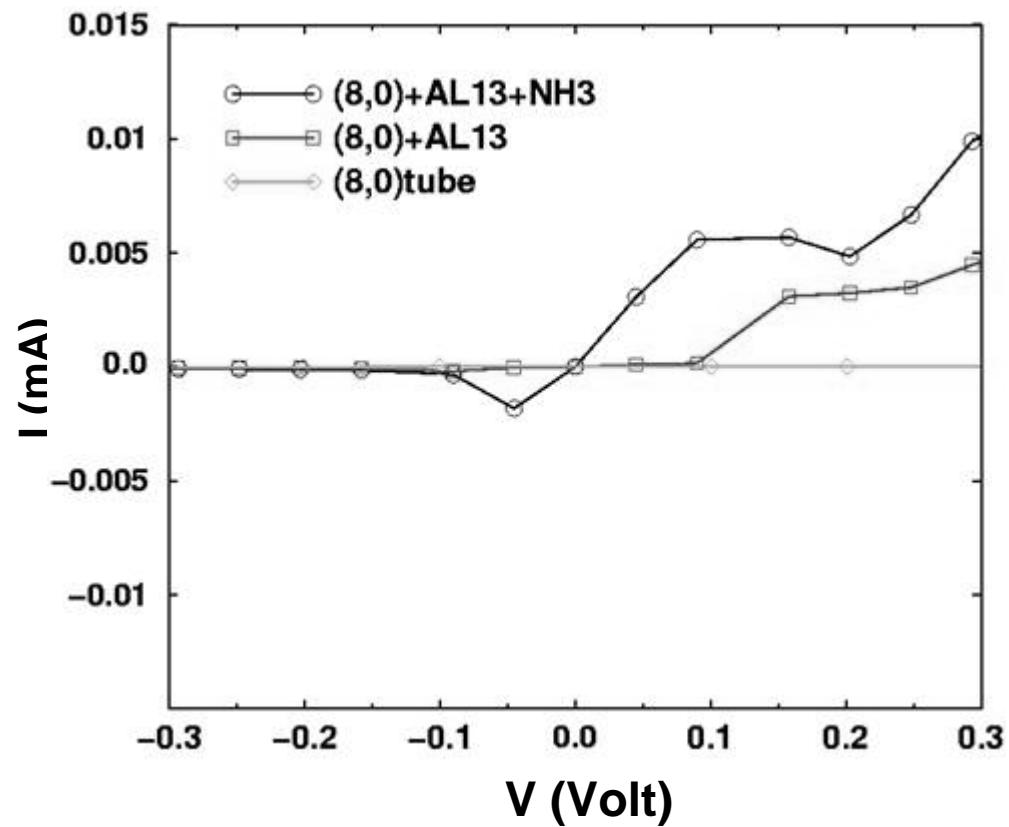
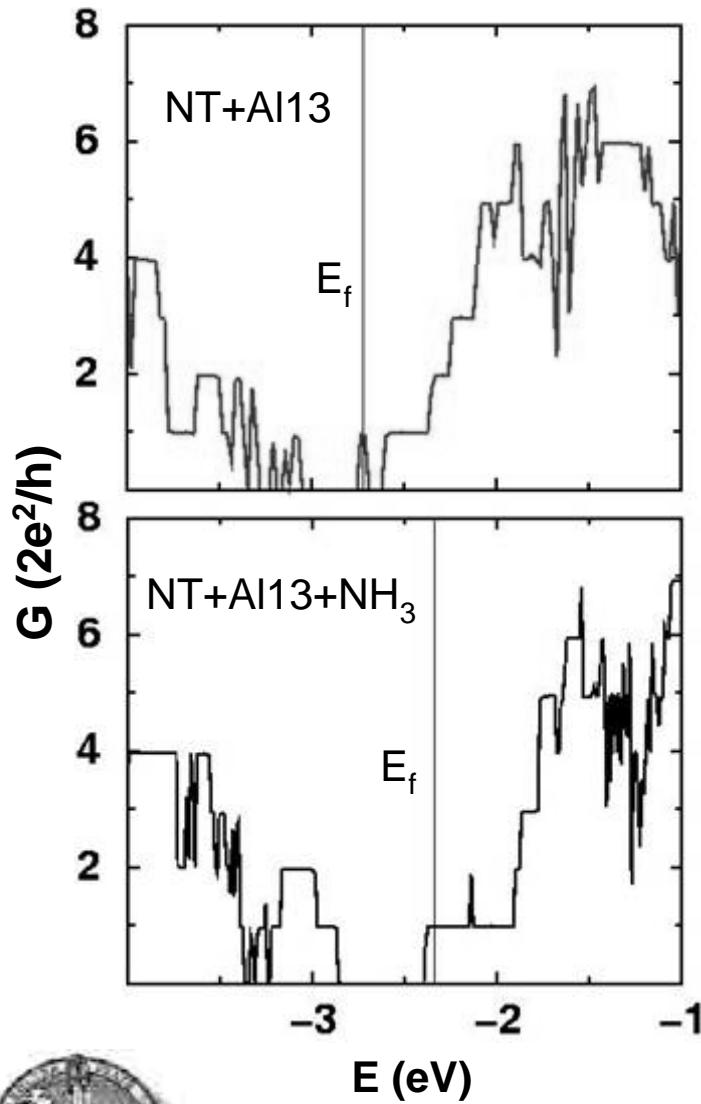
(8,0)tube + Al_{13}
More stable
C-Al bond length: 2.15 Å
Formation energy: -0.7 eV



(8,0)tube + $\text{Al}_{13} + \text{NH}_3$
More stable
C-Al bond length: 2.11 Å
**Formation energy of the
molecule complex: -1.8 eV**



Quantum transport in semiconducting SWNT-metal cluster assembly



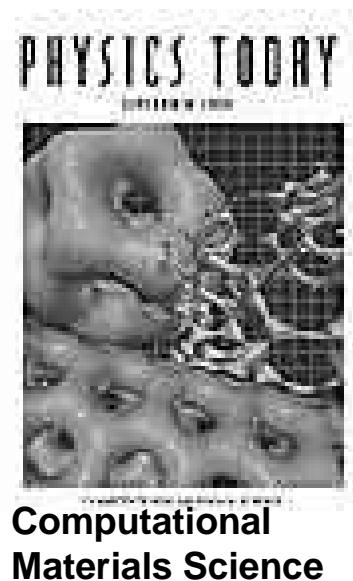
**(8,0)tube-met al cluster assembly becomes
conducting after NH₃ adsorption.
SWNT-met al cluster assembly can be a
good chemical sensor**



Summary

Nanotubes – novel material with truly spectacular properties

- Mechanical properties
 - ❖ The strongest materials known!
 - ❖ The ultimate strength has not yet been reached
- Li/NT high capacity and high throughput batteries
- BN-NTs are excellent pyro- and piezoelectrics
 - ❖ Novel devices: actuators, transducers, etc
- Carbon-based molecular electronics possible and exciting
 - ❖ Strain and molecular sensors
 - ❖ Single electron devices
- Many unique applications possible



Computational
Materials Science